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N-Heterocyclic Carbene-Catalyzed Generation of α , β -Unsaturated Acyl Imidazoliums: Synthesis of Dihydropyranones by their Reaction with Enolates

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N-Heterocyclic carbene (NHC) catalyzed polarity reversal of the carbonyl group is central to a number of organic transformations, most notably the benzoin and Stetter reactions.^{1a,c} More recently conjugated or redox active substrates have allowed access to reactive intermediates useful in reaction discovery.^{1b,d} We have undertaken studies aimed at extending this family to include α,β -unsaturated acyl azoliums (i.e., **2**). While these intermediates were reported by Scheidt in his MnO₂ based oxidative esterification,^{2a} their reactivity at the β -position is yet to be investigated. In comparison to homoenolates, which react as donors at the β -position (eq 1),^{1d,3} the cationic nature of the acyl azolium, in combination with steric shielding, should generate an intermediate ideally suited to act as a conjugate acceptor. In this communication, we report the first conjugate addition reactions involving α,β -unsaturated acyl azoliums (eq 2).

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To investigate the reactivity of α,β -unsaturated acyl azoliums, a method for their generation in the presence of an appropriate nucleophile was required. While a number of approaches could be envisaged, we chose to investigate the use of α,β -unsaturated enol esters (i.e., **5**).⁴ When such substrates are exposed to Lewis basic catalysts,⁵ fragmentation should liberate enolate **3** and the required α,β -unsaturated carboxylate **2**.^{6,7} Conjugate addition into the acyl azolium, providing enolate **6**,⁸ followed by proton transfer and acylation, should then afford pyranone **4**^{9,10} and regenerate the catalyst (Scheme 1).

Investigations began with the reaction of thiamine (**A**) and enol ester **5a** in refluxing toluene (Table 1, entry 1). This readily available NHC fragmented enol ester **5a**, however, rather than affording **4a** *C*-acylation provided 1,3-diketone **7**.¹¹ When 100 mol % of the more nucleophilic tetramethyl carbene **B1** was reacted with enol ester **5a**, pyranone **4a** was obtained in 41% yield, accompanied by 22% of diketone **7** (Table 1, entry 2). Unfortunately, when carbene **B1** was used substoichiometrically no more than a trace of dihydropyranone **4a** was observed (Table 1, entry 3). To our delight, when catalysts derived from IMes•HCl **C1** and IDip•HCl **C2**, recognized in both homoenolate and transesterification catalysis,^{3,6} were used, pyranone **4a** formed in high yield with good catalytic loading (Table 1, entries 4 and 5).

Finally the suitability of imidazolium derived catalyst **B2** and triazolium precatalyst **D** was investigated. Unfortunately neither proved suitable for this reaction (Table 1, entries 6 and 7).¹²

Scheme 1. Proposed Catalytic Pathway for the Formation of 4



Table 1. Selected Optimization Studies



^{*a*} Isolated yield following flash column chromatography. ^{*b*} Catalyst **B2** generated by deprotonation of the corresponding imidazolium chloride.

The reactions' scope was investigated across a range of α,β unsaturated acceptors and enolates (Chart 1). Aromatic, heteroaromatic, and aliphatic α,β -unsaturated esters all reacted readily to give the corresponding pyranones (**4a**-**d**). β -Disubstituted acyl azoliums were found to be good substrates, allowing efficient construction of quaternary carbon centers (**4d**, **4f**, **4m**). Ketone and aldehyde enolates derived from cyclic, acyclic, aromatic, and 1,3diketones were investigated in the transformation providing good yields of the expected dihydropyranones. In the case of 1,3diketones, the reaction preceded readily at 0 °C, even with a β -disubstituted acyl azolium intermediate (**4m**).

While the generation of acyl azoliums and enolates from enol esters proved useful, these should also be accessible from α , β -unsaturated acid fluorides **8** and TMS enol ethers **9** (eq 5). In this



^{*a*} Isolated yield. ^{*b*} Performed on 5 mmol of **5a**. ^{*c*} Ratio determined by ¹H NMR spectroscopy. ^{*d*} 20 mol % **C2**, 40 mol % KO'Bu at -78 °C then warmed to rt. ^{*e*} 20 mol % **C1**, 40 mol % KO'Bu at 0 °C.

scenario addition of the NHC into the acid fluoride would instigate deprotection of the enol ether, unmasking the enolate.¹³ When various α , β -unsaturated acid fluorides were reacted with TMS enol ethers in the presence of 20 mol % **C2** and 40 mol % KO'Bu, the dihydropyranones formed in good yield (Chart 2).

Chart 2. Scope of Intermolecular NHC Catalyzed Reaction^a



^{*a*} Isolated yield. ^{*b*} Ratio determined by ¹H NMR spectroscopy. ^{*c*} 10 mol % C1, 20 mol % KO'Bu at 0 °C.

Recently, highly enantioselective Stetter and benzoin reactions have been achieved using chiral NHCs.^{1a} When catalysts derived from either chiral triazolium¹⁴ or imidazolium¹⁵ salts were exposed to enol ester **5a**, pyranone **4a** failed to form. However, when the reaction was attempted with enol ester **5k** and the catalyst derived from triazolium **E**, pyranone **4k**¹⁶ formed in good yield and 50% ee (eq 6). While this level of induction is modest, it is comparable to the state-of-the-art for reactions in which bond formation β to the carbonyl group occurs (eq 1).^{2b,3a,12a}



In summary, we have achieved the first conjugate addition reactions into α , β -unsaturated acyl azoliums. A range of conjugate acceptors, included those containing β -disubstitution, reacted smoothly with a variety of enolates to afford the expected dihydropyranones in good yields. This transformation complements NHC-catalyzed homoenolate chemistry that involve β -donation (eq 1).^{1d,3} An enantioselective variant of this reaction has been achieved with triazolium **E**. Exploration of the utility of this new type of azolium intermediate and the application of pyranones, known medicinal agents,¹⁷ and useful building blocks in organic synthesis^{10a,b} is ongoing.

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Supporting Information Available: Characterization data, NMR spectra, and detailed experimental procedures. This material is available free of charge via the Internet at http://pubs.acs.org.

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